

# CHARACTERIZATION OF PALM KERNEL SHELL POWDER FOR USE IN POLYMER MATRIX COMPOSITES

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**Abstract--** The use of biomaterials as fillers/reinforcing agents in thermoplastics /thermosets materials is increasing nowadays due to economic reasons and environmental awareness. This work was aimed at characterising the powdered palm kernel shell for use in composite materials formulation. Properties such as powder porosity, hydration capacity, moisture sorption, particle size distribution, bulk density, tapped density, powder flow, pH of powder dispersion as well as differential scanning calorimetry thermal (DSC), X-ray diffraction (X-RD) and scanning electron microscopy (SEM) were evaluated. The following results were obtained; pH was found to be  $4 \pm 0.0$ , average true density, powder porosity and hydration capacity were;  $1.58 \pm 0.07 \text{g/cm}^3$ ,  $6.76 \pm 0.42\%$ , and  $150.08 \pm 76.91\%$ , respectively. The average values of the moisture content and moisture sorption were;  $11.16 \pm 0.16\%$  and  $2 \pm 0.54\%$ , respectively. The average angle of repose for the palm kernel shell (PKS) was found to be  $34.09 \pm 4.77^\circ$ . The scanning electron microscopy showed that PKS powder had spherical shape. Based on EDX analysis PKS has a high carbon content of about 63 wt %. The XRD pattern for the PKS powder showed that the PKS powder was more of amorphous material with small amount of micro crystalline materials. PKS is therefore suggested as filler to be used in polymer composite production.

**Index Term--** palm kernel shell powder, physico-chemical properties, thermal characterizations.

## INTRODUCTION

The use of biomaterials in general and agro-waste in particular is a subject of great interest nowadays not only from the technological and scientific points of view, but also socially, and economically, in terms of employment, cost and environmental issues. Nigeria is endowed with a lot of mineral and agro-based resources that could be used in the development of environmental- friendly composite materials such as Eco-pad used in modern vehicle braking systems. As of 2009, Indonesia was the largest producer of palm oil, Dagwa,

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surpassing Malaysia in 2006, producing more than 20.9 million tonnes. Food Agriculture Organisation (FAO) data showed production increased by over 400% between 1994–2004, to over 8.66 million metric tonnes. In 2008, Malaysia produced 17.7 million tonnes of palm oil on 4,500,000 hectares of land, and was the second largest producer of palm oil, employing more than 570,000 people. Malaysia is the world's second largest exporter of palm oil. As of 2011, Nigeria was the third-largest producer, with more than 2.5 million hectares ( $6.2 \times 10^6$  acres) under cultivation. Until 1934, Nigeria had been the world's largest producer. Both small- and large-scale producers participated in the industry[1].

In the light of the above, large quantities of cracked palm kernel shell (PKS) are therefore generated by the producers. The PKS are obtained after extraction of the palm oil the nuts are broken and the kernels are removed with the shells mostly left as waste. The PKS are hard stony endocarps that surround the kernel and the shells come in different shapes and sizes ([2]. These shells are mainly of two types the “Dura” and “Tenera”. The Tenera is a hybrid which has specially been developed to yield high oil content and it has a thin shell thickness compared to Dura type [3]. There are several efforts being made towards the utilization of the PKS. Some of the areas where palm kernel shell are used or are being considered for use include: automobile disk brake pad [4, 5], carbon activation for water purification, concrete ingredient in building industry, fuel for heat generation [6], thermal insulator [7], etc. The shell is made up of 33% charcoal, 45% pyroligneous liquor and 21% combustible gas as reported [6]. The use of materials such as rice husk, bagasse, palm kernel shell powder, etc. as fillers and/ or reinforcement agents in polymers and composite materials manufacture such as in brake pads have been reported by several authors [4,5,8,9]. However, some of the detailed biomaterials properties are scarcely found in literatures.

Therefore, in this work we have attempted to characterize palm kernel shell powder towards generating a data base on PKS powder properties so that it could be exploited for industrial utilization.

## EXPERIMENTAL

### Materials

Palm kernel shells were obtained from the Nigeria Institute For Oil Research(NIFOR) mill, Benin city.

Methods

Preparation of specimen

The cracked PKS were thoroughly washed in clean water with small quantity of detergent to remove residual oil on the shells. Thereafter, the cracked shells were sun dried for 72 hours. Three kilogram quantity of the shells was pulverized using an engine mill, the desired particle size was obtained after 3 passing through the machine.

Material Properties:

pH

One gram and 10g of powder was dispersed in 100ml distilled water with continuous stirring for 1 hour. The pH was determined with a corning pH meter 215 (USA) P<sup>H+</sup> meter.

Particle Size Analysis

The particle size distribution was determined using 1000g quantity of the pulverized kernel which was placed in the sine shaker holding a set of sieves. The shaker was set to vibrate for 15 min at amplitude of 150.

Bulk and Tapped Densities

Powder samples of 10g quantity each was placed in a 50ml clean, dry measuring cylinder and the volume V<sub>0</sub> occupied by each of the samples without tapping was determined. After 1050 taps with the stamp volumeter the volumes were recorded. The bulk and tapped densities was calculated as the ratio of weight to volume (V<sub>0</sub> and V<sub>1050</sub> respectively).

True density

The true densities (ρ<sub>t</sub>) of the PKS were determined by the liquid displacement method using xylene as the immersion fluid and computed using equation (1) as reported[10]:

$$\rho_t = \frac{w}{[(a+w)-b]} \times SG \tag{1}$$

Where w is the weight of powder, SG is specific gravity of liquid (xylene) (=0.8802), a, is weight of pycnometer + xylene and b is weight of pycnometer + xylene + PKS powder.

The weight of pycnometer filled with xylene was determined and 1 g of the powdered kernel shell was transferred into the xylene filled pycnometer and the weight determined. The true density was then evaluated.

Powder Flow

The flow property of the powdered PKS were determined by direct methods by measuring the flow rates: Ten grams (10g) weight (w) of the PKS powder was placed in the flow apparatus and allowed to flow through the funnel orifice. The time taken (t) for the PKS powder to flow through the orifice was noted. Hence, the flow rate was calculated as mass of powder divided by the time taken (w/t).

Compressibility Index (CI) was determined by imputing the values of the bulk and tapped densities according to equation 2.

$$CI = \frac{(tapped\ density - bulk\ density)}{tapped\ density} \times 100\% \tag{2}$$

The static angle of repose (α), was measured according to the fixed funnel and free standing cone method and the tangent of the angle of repose calculated using the equation 3.

$$\tan \alpha = \frac{2h}{D} \tag{3}$$

Where h is the height of the heap of PKS powder and D is the diameter of the base of the heap of powder.

Powder Porosity

This was calculated from the values of true and bulk densities using the formula:

$$e = \left[ 1 - \frac{B_b}{\rho_t} \right] \times 100\% \tag{4}$$

Where B<sub>b</sub>, is the bulk density, ρ<sub>t</sub> is the true density and e is the porosity.

Hydration Capacity

5 g (W<sub>d</sub>) quantities of dry powder of the differentiated particles sizes of the powdered palm kernel shells were placed in 25 ml capacity beaker. 20 ml of distilled water was then added and the content stirred with a glass stirring rod. The dispersion was then allowed to stand for 24 h. Excess water was removed by decanting and upturning until all water has drained. The weight of hydrated mass (W<sub>h</sub>) was then determined. Therefore, the hydration capacity, (Q) was determined using equation (5) as follows:

$$\%Hydration, Q = \frac{(w_h - w_d)}{w_d} \times 100\% \tag{5}$$

Moisture Content

5 g (W<sub>1</sub>) quantities of the different particle sizes of the powdered palm kernel shells were placed in an aluminium foil and placed in an (Genlabs N30C, England) oven set at 100 °C. The foils and there content were weighed every 3 h until a constant weight (W<sub>2</sub>) was established. Moisture content was determined using equation (6):

$$\%Moisture\ content = \frac{(w_2 - w_1)}{w_1} \times 100\% \tag{6}$$

Differential Scanning Calorimetry (DSC)

DSC analysis was carried out on the palm kernel shell to determine the Glass transition (T<sub>g</sub>) and Crystallization (T<sub>c</sub>) temperatures and other thermal behaviour of the PKS powder from the thermograms. The studies were carried out using a

DSC machine (DSC 204F1-Phoenix NETZSCH, Germany) equipped with a thermal analysis system. Indium (156.8 °C) was used as the internal standard. Approximately 1mg of the powdered palm kernel shell (particle size <150 μm) was placed in an aluminum pan (25μl) and covered with a perforated lid. Dry nitrogen was used as the purge gas (purge 20 ml min<sup>-1</sup>). The probes were heated from 25 to 500 °C at a rate of 10 °Cmin<sup>-1</sup>. The relevant thermodynamic parameters were evaluated with Proteus analysis software.

#### X-ray Diffraction (XRD).

The PKS samples were grinded to powder form. Thereafter, the XRD analysis for the raw palm kernel shell powder was taken to detect the presence of crystalline phase or structure. This was carried out using a PAnalytical X'Pert Pro MPD XRD at room temperature with CuKα radiation.

#### Scanning Electron Microscopy (SEM)

Carl Zeiss MA-10 SEM with EDX was used to examine the morphology of the PKS powders.

### RESULTS AND DISCUSSION

#### pH

The Ph of the PKS powder using a Dispasin of 1% and 10% were found to be  $4 \pm 0.08$  and  $4.1 \pm 0.08$ , respectively.

#### Characterization of particle size

The particle size distribution of the PKS powder which was obtained after grinding is as shown in Fig. 1

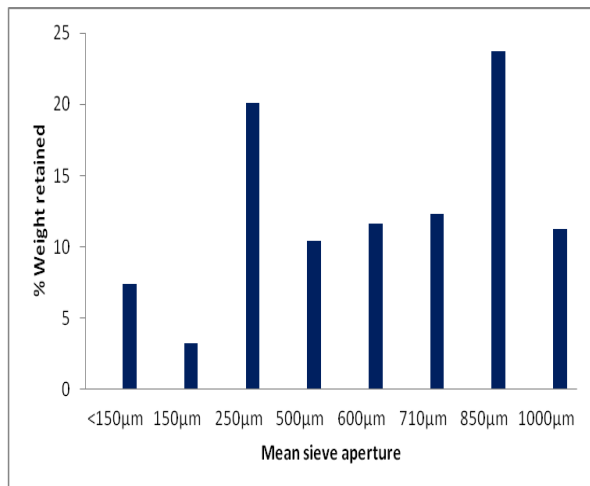


Fig. 1. Particle size distribution of palm kernel shell powder after three passes in the grinding

#### Flow Properties

From Fig. 2a, it was observed that the true density increases with increase in particle sizes of the PKS powder, this had similar behaviour with the densities of sawdust particle sizes as reported by Cardona (2009). For smaller particle sizes, their compressibilities were higher this was as presented in Fig2b because they had more porosity (Fig2c). The smaller particle sizes also had higher hydration capacities as presented in (Fig.2d) this was probably due to the increase in surface area. The average true density, compressibility index, powder

porosity and hydration capacity were;  $1.58 \pm 0.07 \text{g/cm}^3$ ,  $18.58 \pm 5.58$ ,  $6.76 \pm 0.42\%$ , and  $150.08 \pm 76.91\%$ , respectively. From Fig 2e-i it was observed that the smaller the particle sizes the more the moisture sorption and angle of repose, these could be attributed to the increase in surface area and inter-particle friction leading to smaller flow rates, respectively. The moisture content appears to be constant as it varied between  $11.0 \pm 0.15$  to  $11.5 \pm 0.42\%$ . The average values of the following properties: moisture content, moisture sorption, angle of repose, were;  $11.16 \pm 0.16\%$ ,  $2 \pm 0.54\%$ ,  $34.09 \pm 4.77$ , respectively. The angle of repose for the PKS was found to be  $34.09 \pm 4.77$ . From the classification of flow properties by Carr [11] if the angle of repose is between 31 and 35, the material flow is good, whereas materials having values below 30 flow easily and well. The rougher and more irregular the surface of the particles the higher is the angle of repose.

The average compressibility index and Hausner ratio were  $18.58 \pm 5.58$  and 1.2328, respectively, which falls within the range of fair flowability based on the scale of flowability [11], while values above 35 % indicate cohesiveness and Hausners ratios greater than 1.25 indicate poor flow, which would require the addition of a lubricant such a glidant depending on the purpose, to further improve flow [12]. However, it was observed that for smaller particle size (see Fig.2b) the compressibility index was high due to the reason earlier mentioned elsewhere.

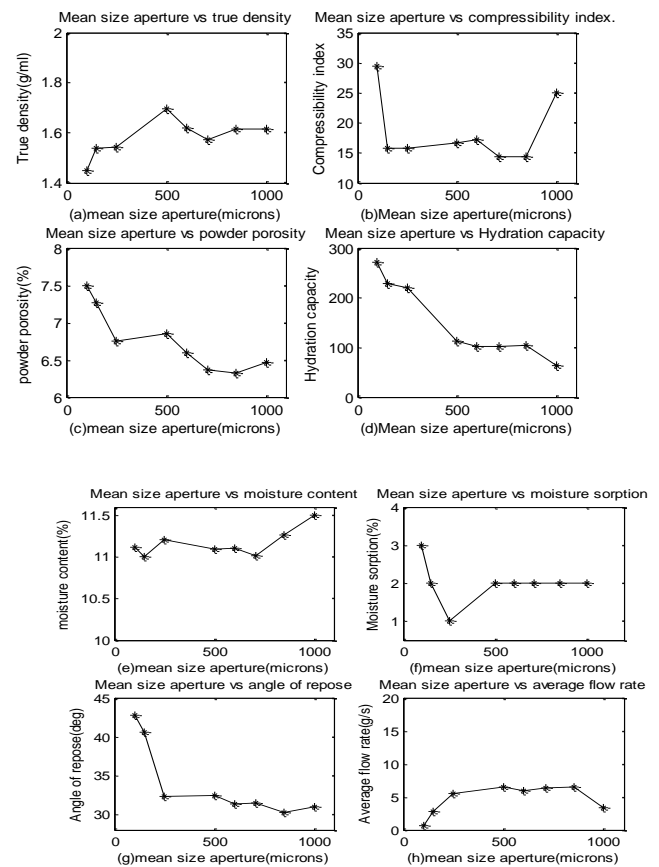


Fig. 2. Powder properties

### Morphological Characterization

Fiber-matrix interface plays a vital role in composite properties. This is because the formation of mechanical bonding at the surface is mainly dependent on the surface topology of the fibers/fillers. Scanning electron microscopic analysis examined the surface topology of the PKS powder and its shape was observed to be spherical. Figure 3 shows the SEM photograph of PKS powder.

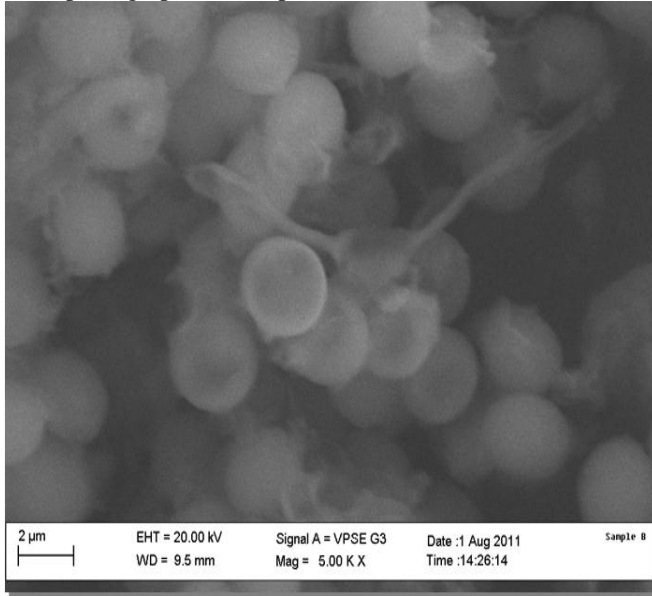


Fig. 3. SEM image of PKS powder

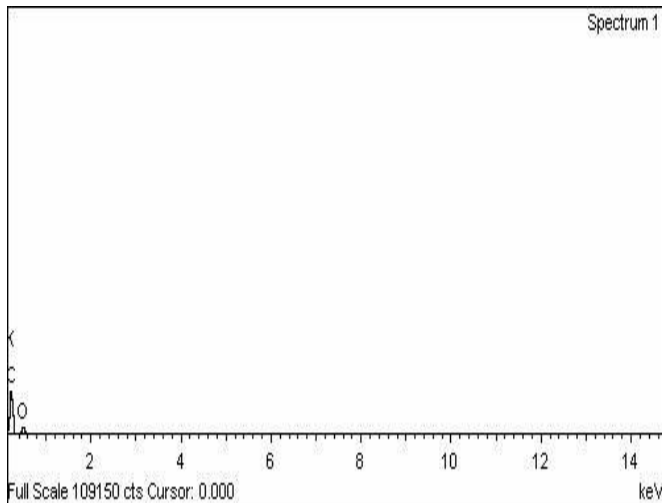


Fig. 4. The EDS spectrum shows the presence of C, O

### DSC analysis

From Fig.5, Thermogram results showed an initial broad endothermic peak at 56°C was observed which could be attributed to the loss of volatile compounds as melting took place or due to removal of absorbed moisture in the PKS powder which is amorphous and partially crystalline. The aforementioned statement is in agreement with the report [13], thus; in all TG curves, the first weight loss step refers to

volatile components of the samples. While, the second phase (around 150°C to 430°C) corresponds to cellulose weight loss and the third part (starting around 430°C) is attributed to lignin thermal decomposition.

In the region between 100 °C -260 °C showed no exothermic or endothermic reactions, which suggest stability. The exothermic hump in the DSC at about 260 °C suggests the commencement of the thermal degradation of hemi-cellulose and the glycosidic linkages of cellulose [14]. Within the temperature ranges the commencement of charring was not noticed which suggest that the degradation of cellulose, leading to the formation of char for PKS was possibly at a temperature greater than 500 °C. This unit was capable of operating up to a maximum temperature of 500 °C. The glass transition temperature (T<sub>g</sub>) could not be observed.

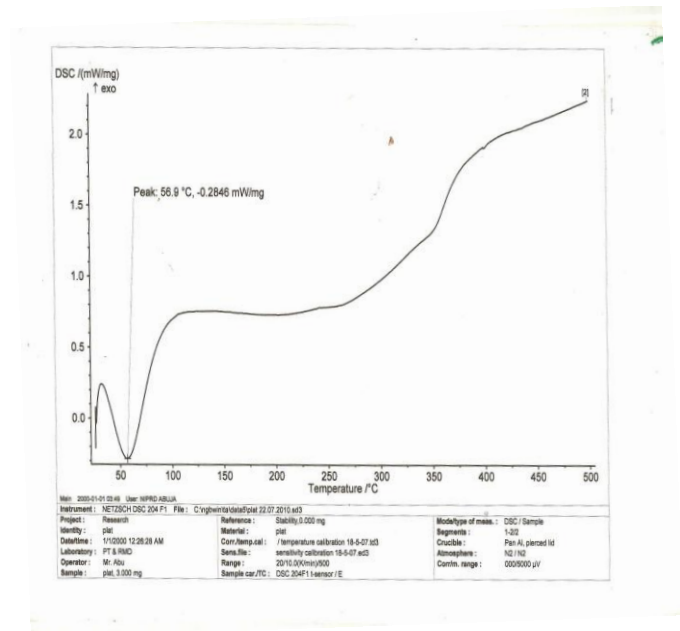


Fig. 5. DSC thermogram

### EDX Analysis

Table 3 presents the elemental weight percent (wt%) obtained from EDX analysis. The raw PKS showed the presence of carbon, oxygen, aluminium, silicon, phosphorous, and potassium with 63.02, 36.04, 0.43, 0.17, 0.17, and 0.17 wt %, respectively. This, to a great extent is in agreement with work reported [15] for raw PKS EDAX analysis which showed the presence of carbon, oxygen, aluminium, and silicon with 46.18, 45.08, 3.47 and 5.27 wt %, respectively. Although, the major elements present were quite similar, the differences observed in the elemental composition could be attributed the source, soil nature, and variety of the palm kernel shells. However, carbon and oxygen were not included in the elemental composition reported [5].



TABLE III  
Energy dispersive x-ray

Element	Weight (%)	Atomic %
C K	63.02	69.67
O K	36.04	29.91
Al K	0.43	0.21
Si K	0.17	0.08
P K	0.17	0.07
K K	0.17	0.06
Total	100.00	100.00

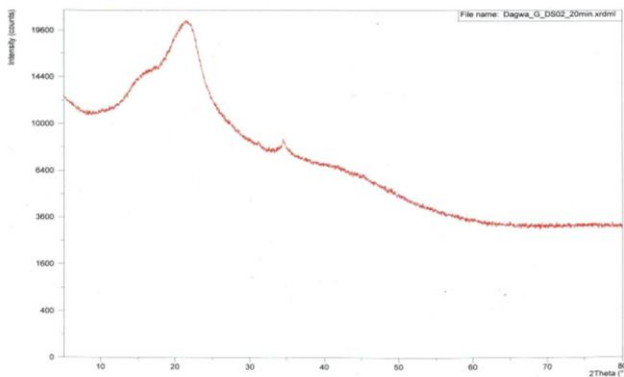


Fig. 6. XRD pattern of PKS powder

Fig. 6. shows the XRD pattern for the PKS powder. This broad peak shape of the diffractogram suggests that the PKS powder was in amorphous state while very small sharp peaks suggests small amount of micro crystalline materials could be present.

Fig. 7 shows a typical performance of PKS based automobile disk brake pad compared with the commercial disk brake pad as reported [4]. This shows that the PKS powder is potentially good filler for use in polymer composite manufacture as the correlation coefficient between the stopping distance of the commercial and prototype (Laboratory) brake pads was 0.9935.

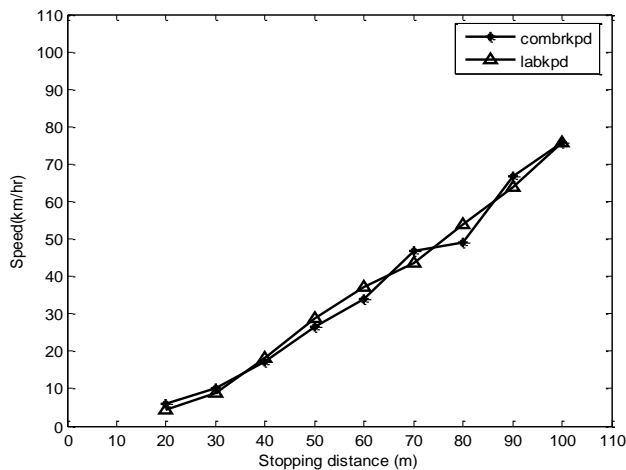


Fig. 7. Effect of vehicle speed on stopping distance (source: Dagwa and Ibhadoe[4])

## CONCLUSIONS

From this study the following conclusions were drawn: Some of the properties possessed by the PKS favour its utilisation as a friction material. Especially, the average true density ( $1.58 \pm 0.07 \text{g/cm}^3$ ), powder porosity ( $6.76 \pm 0.42\%$ ), moisture content ( $11.16 \pm 0.16\%$ ) and moisture sorption ( $2 \pm 0.54\%$ ). The average angle of repose for the PKS was found to be  $34.09 \pm 4.77^\circ$  which suggests that during production, the powder material will flow with ease. From the SEM image, it was observed that PKS powder had spherical shape. The aforementioned properties of the PKS and the close performance level between the PKS based brake pad and the commercial brake pad (complex composite) suggests that PKS powder is potentially good filler which could be used in the manufacture of polymer composite. Further research is recommended on the study of environmental effect on the properties of PKS based composite.

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